

Bipodal acylthiourea ligands as building blocks for Bi-, Tetra-, and polynuclear oxorhenium(V) complexes

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Abstract: Reactions of $(\text{NBu}_4)[\text{ReOCl}_4]$ and 3,3,3', 3'-tetraalkyl-1,1'-isophthaloylbis(thioureas), $\text{H}_2\text{phth}(\text{R}_2\text{tu})_2$ where $\text{R} = \text{Et}, i\text{-Bu}$, in hot MeOH with the addition of Et_3N give red products of the composition $[\text{ReO}(\text{OMe})\{\text{phth}(\text{R}_2\text{tu})_2\}]_2$ (8a, $\text{R} = \text{Et}$; 8b, $\text{R} = i\text{-Bu}$). X-ray structures of 8 reveal symmetric binuclear complexes containing two almost coplanar organic ligands, each of which coordinates to two rhenium centers via the two bidentate-O,S moieties. The octahedral coordination spheres of the rhenium atoms are completed by each one oxo and one methoxido ligand which are directed perpendicular to the plane defined by the $\{\text{phth}(\text{R}_2\text{tu})_2\}^{2-}$ ligands. While in 8a, both methoxido ligands point to the same side of the described plane and form a syn isomer, the MeO^- ligands in 8b are located at opposite sides and form an anti isomer. Studies in solution show that there exists a reversible equilibrium between the anti and syn isomers. Dimerization/condensation of complexes 8 with the formation of tetranuclear complexes of the composition $[\{\text{ReO}\{\text{phth}(\text{R}_2\text{tu})_2\}\}_2\text{O}]_2$ (9) and/or polynuclear species is observed in solutions, which do not contain MeOH. © 2010 American Chemical Society.

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